Synthesis of Optically Active 2- and 3- Indolylglycine Derivatives and their Oxygen Analogues

Koushik Goswami, Indranil Duttagupta, and Surajit Sinha*

Department of Organic Chemistry, Indian Association for the Cultivati[on](#page-4-0) of Science, Jadavpur, Kolkata 700 032, India

S Supporting Information

[AB](#page-4-0)STRACT: [2-Indolylglyc](#page-4-0)ine derivative and its oxygen analogue have been synthesized by Sonogashira coupling followed by cyclization in one pot between 2-iodoheteroarenes and ethynyloxazolidinone where 3-indolylglycine derivative and its oxygen analogue have been synthesized from silylated internal alkyne using Larock's heteroannulation as the key reaction.

Synthesis of indolylglycines is getting importance because of
their presence in various bis-indole alkaloids, namely,
dragonacidina^{1a} and hamacapthing ^{1b} Morogyar, they can be dragmacidins^{1a} and hamacanthins.^{1b} Moreover, they can be used in peptide synthesis to limit conformational flexibility, to enhance enz[ym](#page-4-0)atic stability and [bio](#page-4-0)availability compared to native peptides. Synthesis of optically active 3-indolyglycine derivative has been reported using three distinct approaches: (a) through Friedel–Crafts^{2a–c} reaction taking the advantage of nucleophilic properties at 3-position of indole, (b) via Sharpless asymmetric aminohydr[ox](#page-4-0)y[la](#page-4-0)tion,2d and (c) diastereoselective approach using chiral glyoxalate imines^{2e} or optically active ylide.^{2f} Other [a](#page-4-0)nalogues such as 2-indolylglycine and 2benzofuranyl glycine derivatives are fou[nd](#page-4-0) to be commercially avail[abl](#page-4-0)e; however, to the best of our knowledge, there is no literature report on their synthesis except a racemic route of 2 benzofuranyl glycine.^{2g} Herein, we describe the synthesis of these heteroaryl amino acid derivatives using chiral pool approach.

Retrosynthetic str[ate](#page-4-0)gy A was used for the synthesis of 3 indolylglycine derivative 1a and its "O" analogue 1b using Larock's heteroannulation³ as the key reaction, whereas retrosynthesis B was used for 2-indolylglycine derivative 2a and its "O" analogue 2b by [w](#page-4-0)ay of intermediate 7. In order to avoid epimerization at the α -C^{4b} and undesired intramolecular nucleophilic (O or N) attack to Pd(II)–alkyne complexes,^{4a,c} the masked form of propargylglyci[ne](#page-4-0) derivative was chosen instead of propargylglycine derivative itself (Scheme 1).

Accordingly, acetylated 2-iodoaniline was treated with silylated internal alkyne 3a under Larock's [h](#page-1-0)eteroannulation³ condition to get 2,3-disubstitued indole derivative 4c. Treatment of 4c with TBAF at 0 °C resulted in 5c due to simultaneou[s](#page-4-0) deprotection of sillyl and acetyl groups. Changing the protecting group from acetyl to tosyl enhanced the yield of heteroannulation reaction, and the tosyl group remained unaffected under the condition of sillyl deprotection. Similarly, the oxygenanalogue 5b was synthesized from 2-iodophenol, and a better yield was obtained when TBDMS-substituted internal alkyne 3b was used instead of TMS-protected 3a. Then oxazolidine deprotection 5 of 5a and 5b furnished Boc-protected amino alcohols $6a$ and $6b$, respectively, which after oxidation^{6,5} followed by esterification with diazomethane afforded the

corresponding methyl esters 1a and 1b (Scheme 2). The internal alkynes (3a or 3b)⁷ were prepared according to the literature procedure from ethynyloxazolidine 12,^{8a} whi[ch](#page-1-0) in turn was synthesized from [G](#page-4-0)arner's aldehyde^{8b,c} using Bestmann− Ohira^{8d,e} reagent.

For the synthesis of 2-indolylglyci[ne, w](#page-4-0)e initially followed Knig[ht](#page-4-0)'s [9](#page-4-0) iodocyclization method on 2-alkynylaniline 8b and obtained the desired product 9 in poor yield. When compound 8c was s[u](#page-4-0)bjected under Larock's ¹⁰ iodocyclisation condition, an unexpected nucleophilic attack of Boc led to the formation of 10 instead of expected indole deriv[ati](#page-4-0)ve 11. Cyclisation of 8a using palladium catalysis^{4c} resulted in a complex mixture. From this outcome we realized the problem associated with the Nprotecting group [of o](#page-4-0)xazolidine ring (Scheme 3).

In an attempt to solve the problem, acetonide protection was removed from ethynyloxazolidine 12, and th[e](#page-1-0) resulting amino alcohol 13 on teartment with thionyl chloride furnished ethynyloxazolidinone 14^{11} (Scheme 4).

We were pleased to say that coupling of 14 with tosylated 2 iodoaniline under Son[oga](#page-4-0)shira¹² c[on](#page-1-0)dition gave the desired cyclized product 15a in one pot. Protection of 15a with $Boc₂O$ and subsequent carbamate dep[rot](#page-4-0)ection by cesium carbonate¹³ gave Boc-protected amino alcohol 17a. The alcohol was then co[n](#page-4-0)verted to its methyl ester 2a by means of two step oxidation⁶ followed by treatment with diazomethane. Similarly, 2 benzofuranylglycine derivative 2b has been synthesized from [2](#page-4-0) iodophenol using the same protocol as stated for nitrogen analogue (Scheme 5).

In conclusion, the synthesis of optically active 2- and 3 indolylglycine de[ri](#page-2-0)vatives and their oxygen analogues is described. Though the synthesis of 3-indolylglycine was reported by few groups but we are the first to report the synthesis of 1b, 2a and 2b using chiral pool approach. It is anticipated that minor modifications of the starting materials and methods presented here should give various other analogues of these amino acids.

Received: April 11, 2012 Published: July 26, 2012

Scheme 1. Retrosynthetic Pathway

Scheme 2. Synthesis of 3-Indolylglycine Derivative and Its Oxygen Analogue^a

a
Reaction conditions: (a) 3a, Pd(OAc)₂, PPh₃, n-Bu₄Cl, DIPEA, DMF, 90 °C, 12 h; (b) 3b, Pd(OAc)₂, LiCl, Na₂CO₃, DMF, 90 °C, 25 h; (c) TBAF, THF, 0 °C to rt, 2 h; (d) PTSA, MeOH, rt, 2 h; (e) (i) Dess−Martin periodinane, DCM, rt, 30 min, (ii) NaClO₂, NaH₂PO₄, t-BuOH, H₂O, rt, 1.5 h, (iii) CH_2N_2 , ether, (for 1a); (f) (i) 1 M Jones reagent, acetone, 0 °C, (ii) CH_2N_2 ether, (for 1b).

Scheme 3. Attempted Synthesis toward 2-Substituted Indole Derivative

Scheme 4. Synthesis of Propargylcarbamate^a

$$
\begin{array}{c}\n0 \\
\uparrow \uparrow \\
\uparrow \uparrow \\
\downarrow \uparrow\n\end{array}\n\begin{array}{c}\n\text{a} \\
\uparrow \uparrow\n\end{array}\n\begin{array}{c}\n\text{HO} \\
\uparrow \uparrow\n\end{array}\n\begin{array}{c}\n\text{b} \\
\uparrow \uparrow\n\end{array}\n\begin{array}{c}\n0 \\
\uparrow \uparrow\n\end{array}\n\begin{array}{c}\n\text{a} \\
\uparrow \downarrow\n\end{array}
$$

^aReaction conditions: (a) PTSA, MeOH, rt, 2 h, 61%; (b) $S O Cl₂$ (distilled), THF, rt, 18 h, 80%.

EXPERIMENTAL SECTION

General Information. All reagents were purchased from commercial sources and used without further purification, unless otherwise stated. Petroleum ether (PE) refers to the fraction of petroleum boiling between 60 and 80 °C. THF is the abbreviation of tetrahydrofuran. All reactions were carried out in oven-dried glassware under an argon atmosphere using anhydrous solvents and standard syringe and septum techniques unless otherwise indicated. Organic extracts were dried over anhydrous $Na₂SO₄$ and then filtered prior to

removal of all volatiles under reduced pressure on rotary evaporation. Chromatographic purification of products was accomplished using column chromatography on silica gels (mesh 100−200). Thin-layer chromatography (TLC) was carried out on aluminum sheets, silica gel 60 F254 (layer thickness 0.25 mm). Visualization of the developed chromatogram was performed by UV light and/or phosphomolybdic acid stains. Optical rotations were measured at stated temperature and solvent. Concentration was in gm/100 mL when $[\alpha]_{\rm D}$ was recorded. $^{\rm l}{\rm H}$ and 13C NMR spectra were recorded at 500 and 125 MHz, respectively, using CDCl₃ as solvent. Chemical shifts (δ) are given in ppm relative to the solvent residual peak or TMS as internal standard. The following abbreviations are used for multiplicity of NMR signals: $s =$ singlet, $d =$ doublet, $t = triplet$, $m = multiplet$, $br = broad$. High resolution mass spectra (HRMS) were measured in a QTOF I (quadrupole-hexapole-TOF) mass spectrometer.

Experimental Procedures. (R)-tert-Butyl 2,2-Dimethyl-4-(2- (trimethylsilyl)-1-tosyl-1H-indol-3-yl)oxazolidine-3-carboxylate (4a). To a solution of tosylated 2-iodoaniline (100 mg, 0.27 mmol) and (R)-tert-butyl 2,2-dimethyl-4-(2-(trimethylsilyl)ethynyl)oxazolidine-3Scheme 5. Synthesis of 2-Indolylglycine Derivative and Its Oxygen Analogue^a

^aReaction conditions: (a) 14, Pd(OAc)₂, PPh₃, CuI, DIPA, DMF, 65 °C, 3 h (for 15a), 12 h (for 15b); (b) Boc₂O, THF, Et₃N, DMAP, rt, 16 h (for 16a), 48 h (for 16b); (c) Cs₂CO₃, MeOH, rt, 3 h; (d) (i) Dess–Martin periodinane, DCM, rt, 30 min, (ii) NaClO₂, NaH₂PO₄, t-BuOH, H₂O, rt, 1.5 h, (iii) CH_2N_2 , ether (for 2a); (e) (i) 1 M Jones reagent, acetone, 0 °C, (ii) CH_2N_2 , ether (for 2b).

carboxylate 3a (90 mg, 0.30 mmol) in dry DMF (3 mL) was added $Pd(OAc)₂ (6.0 mg, 0.027 mmol)$ followed by $PPh₃ (28 mg, 0.11 mmol)$, and the reaction vessel was evacuated and flushed with argon three times. To the reaction mixture were added tetrabutyl-ammonium chloride (75 mg, 0.27 mmol) followed by diisopropylethylamine (DIPEA) (0.14 mL, 0.81 mmol) and again flushed with argon. It was heated at 90 °C for overnight. The reaction mixture was cooled to room temperature, and the solvent was removed. The crude residue was subjected for column purification, eluting with petroleum ether−AcOEt (93:7), to afford 4a as waxy solid (94 mg, 64%): ¹H NMR (500 MHz, CDCl₃) δ 7.96 (d, 1H, J = 6.5 Hz), 7.73 (br m, 1H), 7.39 (d, 2H, J = 7.0 Hz), 7.21 (br s, 1H), 7.11 (t, 1H, J = 7.7 Hz), 7.03 (d, 2H, J = 8.0 Hz), 5.32 (br s, 1H), 3.97 (br s, 1H), 3.69 (br s, 1H), 2.24 (s, 3H), 1.78 (s, 3H), 1.61 (s, 3H), 0.93–1.37 (2 br s, 9H), 0.57 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 152.4, 144.2, 140.6, 136.3, 134.4, 130.2, 129.8, 129.6, 129.1, 126.6, 125.4, 123.8, 120.6, 116.2, 95.3, 80.1, 69.0, 55.2, 28.1, 25.5, 21.6, 3.0; HRMS (ESI) $(M + Na)^+$ calculated for $C_{28}H_{38}N_2O_5S$ SiNa⁺ = 565.2168, found 565.2169.

(R)-tert-Butyl 2,2-Dimethyl-4-(1-tosyl-1H-indol-3-yl)oxazolidine-3-carboxylate (5a). To a solution of $4a$ (80 mg, 0.15 mmol) in tetrahydrofuran (3 mL) was added tetrabutyl ammonium fluoride (0.18 mL of 1.0 M in THF, 0.18 mmol) at 0 $^{\circ}$ C, and the mixture was stirred at rt for 1 h. After removal of solvent the crude residue was charged directly into column, eluting with petroleum ether−AcOEt (92:8), to give 5a as white foam (58 mg, 82%): $[\alpha]^{25}$ _D = -65.6 (c 1.79, CHCl₃); IR (neat/ CHCl₃) ν 1697 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.94–7.99 (br m, 1H), 7.73 (br s, 2H), 7.62 (d, 1H, J = 8.0 Hz), 7.43−7.48 (br m, 1H), 7.30 (br s, 1H), 7.20−7.25 (m, 3H), 5.22 (br s, 0.35 H), 5.06 (br s, 0.65 H), 4.26 (dd, 1H, $J = 6.7$, 8.7 Hz), 3.98 (d, 1H, $J = 8.0$ Hz), 2.32 (s, 3H), 1.17−1.78 (m, 15H); ¹³C NMR (125 MHz, CDCl₃) δ 152.1, 145.0, 135.5, 130.0, 128.7, 126.9, 124.8, 124.5, 124.0, 123.3, 119.8, 113.9, 94.7, 94.2*, 80.5, 80.1*, 68.6, 68.2*, 53.9, 28.5, 28.4, 27.4, 26.5*, 24.6, 23.5*, 21.6 (*Asterisk denotes conformer peaks); HRMS (ESI) $(M + Na)^+$ calculated for $C_{25}H_{30}N_2O_5SNa^+ = 493.1773$, found 493.1774.

(R)-tert-Butyl 4-(Benzofuran-3-yl)-2,2-dimethyloxazolidine-3-carboxylate (5b). To a solution of 2-iodophenol (109 mg, 0.495 mmol) and (R)-tert-butyl 4-(2-(tert-butyldimethylsilyl)ethynyl)-2,2-dimethyloxazolidine-3-carboxylate 3b (140 mg, 0.413 mmol) in dry DMF (3 mL) were added lithium chloride (17 mg, 0.41 mmol) and sodium carbonate (131 mg, 1.24 mmol). The reaction vessel was evacuated and flushed with argon three times. Then $Pd(OAc)₂$ (10 mg, 0.041 mmol) was added to it. The reaction mixture was again flushed with argon and heated at 90 °C for 25 h. The reaction mixture was cooled to room temperature, and the solvent was removed under reduced pressure. The crude residue was subjected for column purification, eluting with petroleum ether−AcOEt (96:4), to afford a mixture of 4b (major) and 3b as colorless liquid (127 mg).

To a solution of the mixture of 4b and 3b (125 mg, 0.29 mmol with respect to 4b) in tetrahydrofuran (4 mL) was added tetrabutylammonium fluoride (0.28 mL of 1.0 M in THF, 0.28 mmol) at 0 °C, and the mixture was stirred at rt for 1 h. After removal of solvent the crude residue was charged directly into column, eluting with petroleum ether− AcOEt (96:4), to give 5b as white foam (70 mg, 53% after 2 steps): $[\alpha]^{25}$ _D = -61.5 (c 1.17, CHCl₃); IR (neat/CHCl₃) ν 1697 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.76 (br s, 1H), 7.54–7.61 (br m, 1H), 7.48 $(d, 1H, J = 8.0 Hz)$, 7.29–7.30 (m, 1H), 7.23–7.26 (m, 1H), 5.09–5.24

(br m, 1H), 4.27−4.30 (dd, J = 6.7, 8.7 Hz), 4.11 (br s, 1H), 1.26−1.77 $(m, 15H)$; ¹³C NMR (125 MHz, CDCl₃) δ 155.8, 152.2, 143.5, 143.0^{*}, 125.9, 124.5, 122.7, 121.4, 120.3, 111.8, 94.7, 94.2*, 80.6, 80.3*, 68.5, 67.9*, 52.9, 28.5, 27.4, 26.4*, 24.7, 23.6* (*Asterisk denotes conformer peaks); HRMS (ESI) $(M + Na)^+$ calculated for $C_{18}H_{23}NO_4Na^+ =$ 340.1525, found 340.1525.

tert-Butyl (R)-2-Hydroxy-1-(1-tosyl-1H-indol-3-yl)ethylcarbamate **(6a).** To a solution of 5a (50 mg, 0.106 mmol) in methanol (1.5 mL) was added p-toluensulfonic acid (PTSA) monohydrate (10 mg, 0.053 mmol), and the mixture was stirred at room temperature for 2 h. The solution was then neutralized with saturated aqueous $NAHCO₃$, diluted with AcOEt (2 mL), and washed with brine (2×1 mL). The organic phase was dried (Na_2SO_4) and concentrated. The residue was eluted from a column with petroleum ether−AcOEt (7: 3) to give 6a (37 mg, 80%) as white foam: $[\alpha]^{25}$ _D = -36.4 (c 1.33, CHCl₃), lit.^{2d} $[\alpha]^{20}$ _D = +38.6 (c 1.25, CHCl₃ antipode of 6a); IR (neat/CHCl₃) ν 1699, 3392 cm ⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.97 (d, 1H, J = 8.0 [Hz\)](#page-4-0), 7.76 (d, 2H, J = 8.0 Hz), 7.54–7.56 (m, 2H), 7.33 (t, 1H, J = 7.5 Hz), 7.21–7.26 (m, 3H), 5.06−5.10 (m, 2H), 3.97 (s, 2H), 2.34 (s, 3H), 2.25 (br s, 1H), 1.45 (s, 9 H); ¹³C NMR (125 MHz, CDCl₃) δ 156.1, 145.2, 135.4, 135.3, 130.1, 129.2, 127.0, 125.2, 123.8, 123.5, 121.1, 119.9, 113.9, 80.3, 65.1, 49.6, 28.4, 21.6; HRMS (ESI) (M + Na)⁺ calculated for $C_{22}H_{26}N_2O_5SNa^+ = 453.1460$, found 453.1460.

tert-Butyl (R)-1-(Benzofuran-3-yl)-2-hydroxyethylcarbamate (6b). To a solution of 5b (52 mg, 0.16 mmol) in methanol (2.0 mL) was added p-toluenesulfonic acid (PTSA) monohydrate (15 mg, 0.08 mmol), and the mixture was stirred at room temperature for 2.5 h. The solution was then neutralized with saturated aqueous $NAHCO₃$, diluted with AcOEt (3 mL), and washed with brine (2×1.5 mL). The organic phase was dried (Na_2SO_4) and concentrated. The residue was eluted from a column with petroleum ether−AcOEt (8: 2) to give 6b (38 mg, 85%) as white foamy solid: $[\alpha]^{25}$ _D = -20.3 (*c* 3.05, CHCl₃); IR (neat/ CHCl₃) ν 1695, 3319 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.60–7.61 $(m, 2H)$, 7.48 (d, 1H, J = 8.5 Hz), 7.31 (t, 1H, J = 7.5 Hz), 7.24 (t, 1H, J = 7.7), 5.23 (br s, 1H), 5.06 (br s, 1H), 3.98 (m, 2H), 2.72 (br s, 1H), 1.45 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 156.2, 155.6, 142.4, 126.4, 124.9, 123.0, 120.1, 119.3, 111.9, 80.4, 65.2, 49.1, 28.5; HRMS (ESI) (M + Na)⁺ calculated for $C_{15}H_{19}NO_4Na^+$ = 300.1212, found 300.1211.

tert-Butyl (R)-(Methoxycarbonyl)(1-tosyl-1H-indol-3-yl) methylcarbamate (1a). To a solution of amino alcohol 6a (25 mg, 0.058 mmol) in dichloromethane (1 mL) was added solid periodinane (27 mg, 0.064 mmol) portion wise, and the mixture was stirred for 0.5 h. After consumption of starting material (TLC), it was diluted with ether (1.6 mL), and the resulting suspension was added to 1.3 (M) NaOH (0.6 mL). The reaction mixture was stirred for 15 min, and the ether layer was washed with 0.6 mL of 1.3 M NaOH and 2 mL of water. The organic layer was dried, filtered, concentrated in vacuo and used in the next step without any purification.

To a solution of the above crude aldehyde in t-BuOH (0.16 mL) and 2-methyl-2-butene (0.08 mL) was added a solution of $NaClO₂ (20 mg)$ and NaH_2PO_4 (20 mg) in H_2O (0.3 mL) at rt. The reaction mixture was stirred for 1.5 h and diluted with saturated aqueous $NH₄Cl$ solution (0.5) mL). The organic layer was extracted with AcOEt $(3 \times 1.5 \text{ mL})$. The combined organic layers were dried, filtered and concentrated in vacuo to give the crude product, which was purified by silica gel column

The Journal of Organic Chemistry Note

chromatography, eluting with petroleum ether−AcOEt (60:40), to afford the corresponding acid as white foam (15 mg).

To the solution of above acid (15 mg, 0.034 mmol) in ether (1 mL) was added excess ethereal diazomethane at 0 °C, and the mixture was stirred for 10 min. After evaporation of ether it was directly charged into column, eluting with DCM−AcOEt (95:5), to give 1a as a waxy solid (13.5 mg, 51%, after 3 steps from amino alcohol 6a): $[\alpha]^{26}$ _D = -25.7 (c 0.76, CHCl₃); IR (neat/CHCl₃) ν 1714, 1745, 3443 cm⁻¹; ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$ δ 7.95 (d, 1H, J = 8.5 Hz), 7.76 (d, 2H, J = 8.5 Hz), 7.63 (d, 1H, J = 8.0 Hz), 7.57 (s, 1H), 7.33 (t, 1H, J = 7.7 Hz), 7.22–7.26 $(m, 3H)$, 5.58 (d, 1H, J = 7 Hz), 5.43 (d, 1H, J = 6.0 Hz), 3.74 (s, 3H), 2.34 (s, 3H), 1.44 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 171.2, 155.0, 145.3, 135.4, 135.2, 130.1, 128.8, 127.1, 125.3, 124.8, 123.7, 120.1, 118.3, 113.8, 80.6, 53.0, 50.5, 28.4, 21.7; HRMS (ESI) (M + Na)+ calculated for $C_{23}H_{26}N_2O_6SNa^+ = 481.1409$, found 481.1408.

tert-Butyl (R)-(Methoxycarbonyl)(benzofuran-3-yl) methylcarbamate (1b). To a solution of amino alcohol 6b (20 mg) 0.072 mmol) in acetone (1 mL) at 0 °C was added freshly prepared Jones reagent (1 M, 0.22 mL, 0.22 mmol) dropwise under nitrogen. After completion of the reaction (TLC), it was quenched with isopropyl alcohol (0.25 mL) and partitioned with AcOEt (15 mL) and saturated NH4Cl (5 mL). After stirring the solution for 1 h, the aqueous layer was separated and re-extracted with AcOEt (15 mL), and the combined organic layers were dried, filtered, and concentrated in vacuo to about 5 mL in volume. The solution of the crude acid was cooled to 0 °C. Excess ethereal diazomethane was added, and the reaction was stirred for 10 min. The diazomethane was blown off with nitrogen, and the organic layer was washed with aqueous NaHCO₃ (4 mL), saturated NH₄Cl (4 mL), dried, filtered, and concentrated in vacuo to give the crude product, which was purified by column chromatography, eluting with DCM− AcOEt (95:5), to afford 1b (9 mg, 41%) as white foam: $\lbrack \alpha \rbrack^{25}$ = -75.1 (c 0.99, CHCl₃); IR (neat/CHCl₃) ν 1714, 1747, 3363 cm⁻¹; ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$ δ 7.65−7.67 (m, 2H), 7.49 (d, 1H, J = 8.0 Hz), 7.31−7.34 (m, 1H), 7.25−7.28 (m, 1H), 5.60 (d, 1H, J = 7.0 Hz), 5.50 (br s, 1H), 3.76 (s, 3H), 1.45 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 171.2, 155.7, 155.1, 143.3, 125.7, 125.1, 123.2, 120.2, 116.9, 111.9, 80.6, 53.0, 49.6, 28.4; HRMS (ESI) $(M + Na)^+$ calculated for $C_{16}H_{19}NO_5Na^+$ $= 328.1161$, found 328.1161 .

 (R) -4-Ethynyloxazolidin-2-one (14). To a solution of 12 (550 mg, 2.0 mmol) in methanol (12 mL) was added p-toluenesulfonic acid (PTSA) monohydrate (190 mg, 1.0 mmol), and the mixture was stirred at rt for 3 h. The solution was then neutralized with saturated aqueous NaHCO₃, diluted with AcOEt (30 mL), and washed with brine (2 \times 10 mL). The organic phase was dried $(Na₂SO₄)$ and concentrated. The residue was eluted from a column with petroleum ether−AcOEt (7: 3) to give 13 (226 mg, 61%) as colorless oil.

Thionyl chloride (0.43 mL, 6 mmol) was added to the solution of aminoalcohol 13 (226 mg, 1.2 mmol) in 12 mL of dry THF under argon atmosphere, and the reaction mixture was stirred at room temperature for 18 h. Evaporation of the solvent under reduced pressure gave the crude, which was purified by flash chromatography (1:1 ethyl acetate/ petroleum ether) to get pure propargylcarbamate 14 as yellowish solid (106 mg, 80%): $[\alpha]^{26}$ _D = -7.7 (c 1.67, CHCl₃); IR (neat/CHCl₃) ν 3246, 2121, 1761 cm ⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 6.08 (br s, 1H), 4.54−4.61(m, 2H), 4.36−4.40 (m, 1H), 2.48 (s, 1H); HRMS (ESI) (M + H)⁺ calculated for $C_5H_6O_2N^+$ = 112.0399, found 112.0393.

(R)-4-(1-Tosyl-1H-indol-2-yl)oxazolidin-2-one (15a). To a stirred degassed solution of N-tosyl-2-iodoaniline (184 mg, 0.49 mmol) and 14 (50 mg, 0.45 mmol) in diisopropylamine (5.0 mL) and DMF (2.0 mL) under nitrogen were added $Pd(OAc)₂$ (10 mg, 0.045 mmol), $PPh₃$ (47 mg, 0.18 mmol) and copper iodide (8.5 mg, 0.045 mmol), respectively. The yellowish solution was degassed again and heated at 65 °C for 3 h. After solvent evaporation in vacuo, the residue was directly charged into column, eluting with petroleum ether:AcOEt (1:1), to give 15a, 112 mg (70%) as yellowish white foam: $[\alpha]^{26}$ _D = +146.7 (c 2.00, CHCl₃); IR (neat/CHCl₃) ν 1755, 3271 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 8.09 $(d, 1H, J = 8.5 Hz)$, 7.62 $(d, 2H, J = 8.5 Hz)$, 7.48 $(d, 1H, J = 7.5 Hz)$, 7.33 (t, 1H, $J = 8.0$ Hz), 7.21–7.27 (m, 3H), 6.80 (s, 1H), 6.60 (s, 1H), 5.50 (dd, 1H, $J = 4.0$, 8.5 Hz), 4.88 (t, 1H, $J = 8.8$ Hz), 4.48 (dd, 1H, $J =$ 4.5, 9.0 Hz), 2.35 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 160.2, 145.6,

139.9, 137.5, 135.3, 130.3, 128.8, 126.3, 125.3, 124.2, 121.5, 114.5, 109.2, 71.6, 51.2, 21.6; HRMS (ESI) (M + Na)⁺ calculated for $C_{18}H_{16}N_2O_4SNa^+ = 379.0728$, found 379.0727.

(S)-4-(Benzofuran-2-yl)oxazolidin-2-one (15b). To a stirred degassed solution of 2-iodophenol (54 mg, 0.25 mmol) and 14 (25 mg, 0.22 mmol) in diisopropylamine (3.0 mL) under nitrogen were added Pd(OAc)₂ (5 mg, 0.022 mmol), PPh₃ (23 mg, 0.088 mmol) and copper iodide (4 mg, 0.022 mmol), respectively. The yellowish solution was degassed again and heated at 65 °C for 12 h. After solvent evaporation in vacuo the residue was directly charged into column, eluting with petroleum ether: $ACOEt (1:1)$, to give $15b$, 20 mg (44%) as yellowish white solid: $\left[\alpha\right]_{D}^{26} = -12.4$ (c 0.66, CHCl₃); IR (neat/CHCl₃) ν 1707, 1741, 3244 cm $^{-1}$; ¹H NMR (500 MHz, CDCl₃) δ 7.55 (d, 1H, J $= 7.5$ Hz), 7.46 (d, 1H, J = 8.5 Hz), 7.29–7.32 (m, 1H), 7.23–7.26 (m, 1H), 6.73 (s, 1H), 6.22 (s, 1H), 5.11 (dd, 1H, J = 5.5, 9.0 Hz), 4.70–4.74 $(m, 1H)$, 4.54 (dd, 1H, J = 5.7, 8.7 Hz); ¹³C NMR (125 MHz, CDCl₃) δ 159.4, 155.4, 154.2, 127.7, 125.1, 123.4, 121.5, 111.5, 104.6, 69.1, 50.5; HRMS (ESI) $(M + Na)^+$ calculated for $C_{11}H_{10}NO_3^+ = 204.0655$, found 204.0654.

(R)-tert-Butyl 2-Oxo-4-(1-tosyl-1H-indol-2-yl)oxazolidine-3-carboxylate (16a). To a well stirred solution of 15a (110 mg, 0.31 mmol) in dry THF (8 mL) was added $Et₃N$ $(0.06 \text{ mL}, 0.44 \text{ mmol})$, Boc2O (107 mg, 0.49 mmol). After 10 min DMAP (3.8 mg, 0.031 mmol) was added, and the reaction was stirred at room temperature for 16 h. After that the solution was concentrated under reduced pressure, and the residue was purified by column chromatography with petroleum ether:AcOEt (80:20) as eluent to give 16a (119 mg, 84%): $[\alpha]_{D}^{26} =$ +91.5 (c 0.73, CHCl₃); IR (neat/CHCl₃) ν 1718, 1817 cm⁻¹; ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$ δ 8.03 (d, 1H, J = 8.5 Hz), 7.75 (d, 2H, J = 8.5 Hz), 7.48 (d, 1H, J = 8.0 Hz), 7.31 (t, 1H, J = 7.5 Hz), 7.23−7.26 (m, 3H), 6.63 (s, 1H), 6.04 (d, 1H, $J = 7.5$ Hz), 4.71 (t, 1H, $J = 8.8$ Hz), 4.42 (dd, 1H, $\vec{J} = 2.5$, 9.0 Hz), 2.35 (s, 3H), 1.42 (s, 9H); ¹³C NMR (125 MHz, CDCl3) δ 152.0, 149.1, 145.6, 138.4, 137.2, 135.3, 130.3, 128.8, 126.8, 125.4, 124.2, 121.4, 114.4, 108.4, 84.5, 68.6, 54.0, 28.0, 21.7; HRMS (ESI) $(M + Na)^+$ calculated for $C_{23}H_{24}N_2O_6SNa^+ = 479.1253$, found 479.1254.

(S)-tert-Butyl 4-(Benzofuran-2-yl)-2-oxooxazolidine-3-carboxylate (16b). To a stirred solution of 15b (20 mg, 0.098 mmol) in dry THF (2 mL) were added Et₃N $(0.02 \text{ mL}, 0.14 \text{ mmol})$, Boc₂O $(34 \text{ mg},$ 0.15 mmol). After 10 min DMAP (1.2 mg, 0.001 mmol) was added, and the reaction was stirred at room temperature for 48 h. After that the solution was concentrated under reduced pressure, and the residue was purified by column chromatography with petroleum ether:AcOEt (80:20) as eluent to give **16b** (17 mg, 57%): $[\alpha]_{D}^{26} = -102.0$ (*c* 0.97, CHCl₃); IR (neat/CHCl₃) ν 1716, 1801 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.56 (d, 1H, J = 8.0 Hz), 7.48 (d, 1H, J = 8.5 Hz), 7.32 (t, 1H, J $= 7.2$ Hz), $7.24 - 7.27$ (m, 1H), 6.73 (s, 1H), 5.46 (dd, 1H, J = 4.2, 8.7 Hz), 4.61 (t, 1H, J = 8.7 Hz), 4.45 (dd, 1H, J = 4.0, 9.0 Hz), 1.42 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ ; 155.1, 153.1, 151.8, 148.8, 127.7, 125.2, 123.4, 121.5, 111.6, 105.2, 84.6, 66.0, 53.0, 28.0; HRMS (ESI) (M + Na)⁺ calculated for $C_{16}H_{17}NO_5Na^+$ = 326.1004, found 326.1005.

tert-Butyl (R)-2-Hydroxy-1-(1-tosyl-1H-indol-2-yl)ethylcarbamate (17a). To a well stirred solution of $16a$ (100 mg, 0.22 mmol) in dry MeOH (4 mL) was added Cs_2CO_3 (15 mg, 0.044 mmol) in one portion, and the solution was stirred at room temperature for 3 h. Then the solution was neutralized with solid citric acid and concentrated under reduced pressure. The residue was dissolved in AcOEt (15 mL), washed with brine (15 mL), H_2O (15 mL) and dried over Na_2SO_4 . The residue was purified by column chromatography using petroleum ether:AcOEt (7:3) as eluent to give 17a (80 mg, 85%) as waxy solid: $[\alpha]_{D}^{26} = +174.2$ $(c \text{ 1.58, CHCl}_3)$; ¹H NMR (500 MHz, CDCl₃) δ 8.07 (d, 1H, J = 8.5) Hz), 7.87 (d, 2H, J = 7.0 Hz), 7.41 (d, 1H, J = 7.5 Hz), 7.24–7.26 (m, 1H,), 7.17−7.20 (m, 3H), 6.69 (s, 1H), 5.74 (s, 1H), 5.55 (d, 1H, J = 5.5 Hz), 4.03−4.10 (m, 2H), 2.30 (s, 3H), 1.84 (br s, 1H), 1.48 (s, 9H); 13C NMR (125 MHz, CDCl₃) δ 155.6, 145.0, 140.5, 137.7, 134.8, 129.9, 129.6, 127.2, 124.7, 124.0, 120.9, 115.2, 111.0, 80.0, 65.5, 51.6, 28.5, 21.6; HRMS (ESI) $(M + Na)^+$ calculated for $C_{22}H_{26}N_2O_5SNa^+ =$ 453.1460, found 453.1457.

(S)-tert-Butyl 1-(Benzofuran-2-yl)-2-hydroxyethylcarbamate (17b). To a well stirred solution of $16b$ (17 mg, 0.056 mmol) in dry

The Journal of Organic Chemistry Note

MeOH (1 mL) was added Cs_2CO_3 $(3.6 \text{ mg}, 0.011 \text{ mmol})$ in one portion, and the solution was stirred at room temperature for 3 h. Then the solution was neutralized with solid citric acid and concentrated under reduced pressure. The residue was dissolved in AcOEt (4 mL), washed with brine (2 mL) , H₂O (2 mL) and dried over Na₂SO₄. The residue was purified by silica gel chromatography using petroleum ether:AcOEt (7:3) as eluent to give 17b (13 mg, 85%) as white foamy solid: $[\alpha]^{26}_{\text{D}} = -60.9$ (c 1.19, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 7.51 (d, 1H, J = 7.5 Hz), 7.43 (d, 1H, J = 8.0 Hz), 7.19−7.27 (m, 2H), 6.64 (s, 1H), 5.40 (d, 1H, J = 7.5 Hz), 5.00 (br s, 1H), 4.01 (br m, 1H), 3.92−3.94 (m, 1H), 2.48 (br s, 1H), 1.45 (s, 9H); 13C NMR (125 MHz, CDCl3) δ 155.8, 155.4, 155.0, 128.2, 124.3, 123.0, 121.1, 111.3, 104.2, 80.4, 64.3, 51.3, 28.5; HRMS (ESI) (M + Na)⁺ calculated for $C_{15}H_{19}NO_4Na^+ = 300.1212$, found 300.1212.

tert-Butyl (R)-(Methoxycarbonyl)(1-tosyl-1H-indol-2-yl) methylcarbamate (2a). To a solution of amino alcohol 17a (50 mg, 0.116 mmol) in dichloromethane (2 mL) was added solid periodinane (55 mg, 0.13 mmol) portion wise, and the mixture was stirred for 0.5 h. It was then diluted with ether (3.0 mL), and the resulting suspension was added to 1.3 (M) NaOH (1.2 mL). After stirring the mixture for 15 min, ether layer was washed with 1.2 mL of 1.3 M NaOH and 4 mL of water. The organic layer was dried, filtered, concentrated in vacuo and used in the next step without further purification.

To a solution of the aldehyde in t-BuOH (0.32 mL) and 2-methyl-2 butene (0.16 mL) was added a solution of $NaClO₂$ (40 mg) and NaH_2PO_4 (40 mg) in H₂O (0.6 mL) at rt. The reaction mixture was stirred for 1.5 h and diluted with saturated aqueous NH4Cl solution (1 mL). The organic layer was extracted with AcOEt $(3 \times 3 \text{ mL})$. The combined organic layers were dried, filtered and concentrated in vacuo to give the crude product, which was purified by column chromatography, eluting with petroleum ether−AcOEt (60:40), to afford the corresponding acid as white foam (32 mg).

To the solution of above acid (32 mg, 0.072 mmol) in ether (2 mL) was added excess ethereal diazomethane at 0 °C, and the mixture was stirred for 15 min. After evaporation of ether, it was directly charged into column, eluting with DCM−AcOEt (95:5), to obtain 2a as a waxy solid (28 mg, 53%): $[\alpha]^{26}$ _D = -52.2 (c 1.85, CHCl₃); IR (neat/CHCl₃) ν 1708, 1735, 3371 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.86 (d, 1H, J = 9.5 Hz), 7.73 (d, $2H$, $J = 8.5$ Hz), 7.48 (d, $1H$, $J = 7.5$ Hz), $7.19 - 7.26$ (m, 4H), 6.78 (s, 1H), 6.10 (d, 1H, J = 8.5 Hz), 5.85 (br s, 1H), 3.74 (s, 3H), 2.34 (s, 3H), 1.45 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 170.2, 155.2, 145.1, 136.9, 136.6, 135.8, 129.9, 128.8, 127.1, 125.3, 123.8, 121.5, 114.6, 113.3, 80.5, 53.1, 52.7, 28.5, 21.7; HRMS (ESI) (M + Na)+ calculated for $C_{23}H_{26}N_2O_6SNa^+ = 481.1409$, found 481.1409.

(R)-Methyl 2-(Benzofuran-2-yl)-2-(tert-butoxycarbonylamino) acetate $(2b)$. To a solution of amino alcohol $17b$ $(12 \text{ mg}, 0.043)$ mmol) in acetone (0.70 mL) at 0 °C was added freshly prepared Jones reagent (1 M, 0.13 mL, 0.13 mmol) dropwise under nitrogen. After completion of the reaction (TLC), it was quenched with isopropyl alcohol (0.14 mL) and partitioned with AcOEt (10 mL) and saturated $NH₄Cl$ (3.5 mL). After stirring the solution for 1 h, the aqueous layer was separated and re-extracted with AcOEt (10 mL), and the combined organic layers were dried, filtered, and concentrated in vacuo to about 5 mL in volume. The solution of the crude acid was cooled to 0 °C. Excess ethereal diazomethane was added to it, and the reaction was stirred for 10 min. The organic layer was washed with aqueous NaHCO_3 (3 mL), saturated $NH₄Cl$ (3 mL), dried, filtered, and concentrated in vacuo to give the crude product, which was purified by column chromatography, eluting with DCM−AcOEt (95:5), to afford 2b (5.6 mg, 43%) as white foam: [α]²⁶_D = –110.3 (c 0.44, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 7.55 (d, 1H, J = 7.5 Hz), 7.45 (d, 1H, J = 8.0 Hz), 7.29 (t, 1H, J = 7.5 Hz), 7.23 (t, 1H, J = 7.5 Hz), 6.75 (s, 1H), 5.62 (m, 2H), 3.79 (s, 3H), 1.45 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 169.4, 155.1, 155.0, 151.8, 128.0, 124.9, 123.2, 121.5, 111.6, 105.5, 80.8, 53.3, 52.2, 28.4; HRMS (ESI) (M + Na)⁺ calculated for $C_{16}H_{19}NO_5Na^+$ = 328.1161, found 328.1162.

■ ASSOCIATED CONTENT

S Supporting Information

Spectroscopic data as well as copies of the spectra of all new compounds are provided. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR IN[FORMATION](http://pubs.acs.org)

Corresponding Author

*E-mail: ocss5@iacs.res.in.

Notes

The aut[hors declare no co](mailto:ocss5@iacs.res.in)mpeting financial interest.

■ ACKNOWLEDGMENTS

S.S. thanks DST, India, for financial support by a grant [SR/S1/ OC-38/2007]. K.G. is thankful to IACS and I.D. is thankful to CSIR for their fellowships.

■ REFERENCES

(1) (a) Kawasaki, T.; Enoki, H.; Matsumura, K.; Ohyama, M.; Inagawa, M.; Sakamoto, M. Org. Lett. 2000, 2, 3027. (b) Kouko, T.; Matsumura, K.; Kawasaki, T. Tetrahedron 2005, 61, 2309.

(2) (a) Johannsen, M. Chem. Commun. 1999, 2233. (b) Zhao, J. L.; Liu, L.; Zhang, H. B.; Wu, Y. C.; Wang, D.; Chen, Y. J. Synlett 2006, 96. (c) Wanner, M. J.; Hauwert, P.; Schoemaker, H. E.; Gelder, R.; Maarseveen, J. H.; Hiemstra, H. Eur. J. Org. Chem. 2008, 180. (d) Yang, C. G.; Wang, J.; Tang, X. X.; Jiang, B. Tetrahedron: Asymmetry 2002, 13, 383. (e) Lei, F.; Chen, Y. J.; Sui, Y.; Liu, L.; Wang, D. Synlett 2003, 1160. (f) Higuchi, K.; Takei, R.; Kouko, T.; Kawasaki, T. Synlett 2007, 669. (g) Nicos A. Petasis, N. A.; Goodman, A.; Zavialov, I. A. Tetrahedron 1997, 53, 16463.

(3) Larock, R. C.; Yum, E. K.; Refvik, M. D. J. Org. Chem. 1998, 63, 7652.

(4) (a) van Esseveldt, B. C. J.; van Delft, F. L.; Smits, J. M. M.; de Gelder, R.; Schoemaker, H. E.; Rutjes, F. P. J. T. Adv. Synth. Catal. 2004, 346, 823. (b) Ma, C.; Liu, X.; Li, X.; Flippen-Anderson, J.; Yu, S.; Cook, J. M. J. Org. Chem. 2001, 66, 4525. (c) van Esseveldt, B. C. J.; van Delft, F. L.; de Gelder, R.; Rutjes, F. P. J. T. Org. Lett. 2003, 5, 1717.

(5) Dondoni, A.; Giovannini, P. P.; Massi, A. Org. Lett. 2004, 6, 2929. (6) (a) Dess, D. B.; Martin, J. C. J. Org. Chem. 1983, 48, 4156. (b) Hale, K. J.; Manaviazar, S.; George, J. H.; Walters, M. A.; Dalby, S. M. Org. Lett. 2009, 11, 733.

(7) Meffre, P.; Gauzy, L.; Branquet, E.; Durand, P.; Goffic, F. L. Tetrahedron 1996, 52, 11215.

(8) (a) Crisp, G. T.; Jiang, Y. L.; Pullman, P. J.; Savi, C. D. Tetrahedron 1997, 53, 17489. (b) Garner, P.; Park, J. M. Org. Synth. 1991, 70, 18−28. (c) McKillop, A.; Taylor, R. J. K.; Watson, R. J.; Lewis, N. Synthesis 1994, 31. (d) Ohira, S.; Okai, K.; Moritani, T. J. Chem. Soc., Chem. Commun. 1992, 721. (e) Roth, G. J.; Liepold, B.; Müller, S. G.; Bestmann, H. J. Synthesis 2004, 59.

(9) Amjad, M.; Knight, D. W. Tetrahedron Lett. 2004, 45, 539.

(10) Yue, D.; Larock, R. C. Org. Lett. 2004, 6, 1037.

(11) (a) Brummond, K. M.; Yan, B. Synlett 2008, 2303. (b) Benedetti, F.; Berti, F.; Norbedo, S. J. Org. Chem. 2002, 67, 8635.

(12) (a) Sonogashira, K.; Tohda, Y.; Hagihara, N. Tetrahedron Lett. 1975, 16, 4467. (b) Chinchilla, R.; Najera, C. ́ Chem. Rev. 2007, 107, 874. (13) Monache, G.; Giovanni, M. C. D.; Misiti, D.; Zappia, G.

Tetrahedron: Asymmetry 1997, 8, 231.